Dpp 1

- 1. Which of the following are exothermic process?
 - A. Reaction of water with quicklime
 - B. Dilution of acid
 - C. Evaporation of water
 - D. Sublimation of camphor
 - a) A B
 - b) A D
 - c) BC
 - d) B D
- 2. $MnO_2 + 4HCl \rightarrow MnCl_2 + 2H_2O + Cl_2$

Identify substance which is oxidised

- a) MnCl₂
- b) HCl2HI
- c) H₂O
- d) MnO₂
- 3. Select the oxidising agent among the following
 - $H_2S + I_2 \longrightarrow 2HI + S$
 - a) H₂S
 - b) HI
 - **c)** I₂
 - d) S

4. What type of reaction occurs when current is passed through water ?

- a) Displacement
- b) Combination
- c) Decomposition
- d) Double displacement
- 5. Substance added to food containing oils and fats is
 - a) Oxidant
 - b) Reductant
 - c) Antioxidant
 - d) Antireductant
- 6. Formula of calcium nitride is
 - a) Ca₃N₂
 - b) Ca₂N₃
 - c) CaN₃
 - d) CaN₂
- 7. $Fe_2O_3 + 2Al \longrightarrow Al_2O_3 + 2Fe$
 - a) Combination
 - b) Displacement
 - c) Decomposition
 - d) Double displacement
- 8. Ferrous sulphate on heating produces
 - a) Ferric oxide
 - b) Carbon dioxide

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- c) Oxygen
- d) Water
- 9. Which of following is combustion reaction
 - a) Boiling of water
 - b) Burning of petrol
 - c) Melting of wax
 - d) None
- 10. In oxidation reaction oxidation number....
 - a) Increases
 - b) Decreases
 - c) No effect
 - d) other
- 11 Which of the following reaction is endothermic?

 - (1) C + O2 (2) CaCO3 (3) CH4 + 2O2 CO2 CaO + CO2 CO2 + 2H2O
 - (4) CaO + H2O → Ca(OH)2
- 12 Which of the following reaction is based on activity series of metals?
 - (1) Decomposition reaction
 - (2) Displacement reaction
 - (3) Double displacement reaction
 - (4) Synthesis reaction
- 13 Which of the following reaction is based on activity series of metals?
 - (1) Decomposition reaction
 - (2) Displacement reaction
 - (3) combination reaction
 - (4) Double displacement reaction